16/11/2023, 13:55

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Research Article

Dissociation Kinetics of Gaseous Ozone in Onion (*Allium Cepa* L.) Bulbs

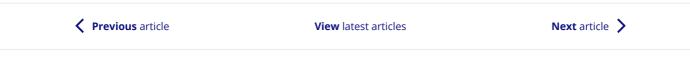
Pramod S. Shelake, Debabandya Mohapatra Solo, Saroj kumar Giri & Subir kumar Chakraborty Received 05 Sep 2022, Accepted 08 Mar 2023, Published online: 04 Apr 2023



ABSTRACT

Ozone decomposition kinetics varies significantly in the presence of biological commodity. This variation depends on many factors but primarily on the active sites in agricultural commodities and storage environments. Ozone decomposition kinetics in onion bulbs at temperatures (2, 10, and 18°C) and relative humidity (RH) (35 & 85%) for different ozone concentrations (100, 200, 300 ppm) and number of exposures (1, 2, 3) were investigated. Zero-order, first-order, and second-order kinetic models were fitted to the ozone decomposition data with respect to time. It was found that the decomposition kinetics follows a first-order reaction and the decomposition rate constant (k_d) varies from $1.712 \times 10^{-3} s^{-1} - 5.181 \times 10^{-3} s^{-1}$ in the presence of onion. The half-life period ($t_{1/2}$) was found to increase with concentration and number of exposures and decrease with temperature and relative humidity. The activation energy was found to vary between 1.840 and 10.216 kJ mole⁻¹ and 6.271–12.461 kJ mole⁻¹ for the relative humidity of 85% and 35%, respectively.

Q KEYWORDS: Activation energy Decomposition Gaseous ozone Onion Temperature



Acknowledgments

16/11/2023, 13:55

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of Science and Technology (DST), Govt. of India, for carrying out the Ph.D. research work at ICAR-CIAE, the outreach program of PG school IARI.

Disclosure statement

No potential conflict of interest was reported by the authors.

Supplementary material

Supplemental data for this article can be accessed online at https://doi.org/10.1080/01919512.2023.2195437

Additional information

Funding

The work was supported by the Department of Science and Technology India [DST-Inspire fellowship (IF180395)].

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